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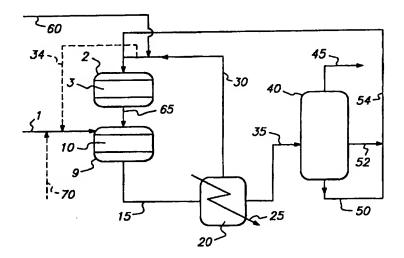
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(54) Title: PROCESS FOR REVERSE STAGING IN HYDROPROCESSING REACTOR SYSTEMS



#### (57) Abstract

A hydrocarbon feed (1) is passed to a denitrification and desulfurization zone (10); passing said denitrification and desulfurization zone effluent to a purification/cooling zone (20) for removal of NH<sub>3</sub> and H<sub>2</sub>S and cooling, and recovering from said purification/cooling zone a hydrogen/light hydrocarbon stream (30) and a liquid stream containing dissolved gases (35); passing said liquid stream containing dissolved gases to a separation zone (40) and recovering a light product (45), a liquid bottoms (50), and at least one side-cut product therefrom (52); passing said liquid bottoms (50) and said side-cut (52) product and said hydrogen/light hydrocarbon stream from step (30) (b) to a hydrocracking or a hydrotreating zone; passing said hydrocracking or hydrotreating zone effluent (65) to said denitrification and desulfurization zone (10).

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1	PROCESS FOR REVERSE STAGING IN
2	HYDROPROCESSING REACTOR SYSTEMS
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4	I. FIELD OF THE INVENTION
5	
6	The present invention relates to the field of hydroprocessing. In particular,
7	the present invention relates to hydroprocessing to obtain high conversions,
8	product selectivity and selective hydrotreating of specific boiling range
9	products.
10	
11	II. BACKGROUND OF THE INVENTION
12	
13	There are two conventional approaches in the petroleum
14	hydroprocessing/hydrotreating art to obtain high conversions. "High
15	conversions" includes sulfur removal, nitrogen removal, hydrocracking,
16	ramsbottom carbon reduction, and the like. The two conventional processes
17	include (a) a long residence time or low space velocity reactors, or (b) a
18	separate reactor loop for the high conversion step after feed impurities are
19	reduced in an initial reactor loop.
20	
21	The second approach using separate reactor loop is effective. This is
22	because the eliminated feed impurity byproducts such as H <sub>2</sub> S, NH <sub>3</sub> , are not
23	present in the typically high concentrations that exist in the first reaction loop.
24	There presence in high concentrations would tend to inhibit reaction rates in
25	the second reaction loop.
26	
27	There exists some conventional approaches in the art for obtaining good
28	product selectivity. "Selectivity" includes obtaining a preferential yield of
29	certain boiling range materials. These conventional processes include
30	(a) recycling the undesirable products for reproc ssing with the fr sh feed, or

1	(b) reprocessing the undesirable products in a separate reaction loop.
2	Typical approaches in the art to selective hydrotreating of specific boiling
3	range products include (a) overtreating of the entire feed to the point where
4	the most difficult product specification is met, or (b) treating of the whole feed
5	to a lesser extent followed by a separate hydrotreating of particular product
6	cuts to meet the most difficult specifications.
7	
8	It would be desirable to have a hydroprocessing process which achieved
9	higher conversion or deeper treating processing while avoiding the
10	drawbacks of known processes.
11	
12	III. SUMMARY OF THE INVENTION
13	
14	The present invention serves to accomplish these objectives in a single
15	reaction loop including lower costs than multiple loops, while maintaining the
16	advantages of a multiple loop system including higher reaction rates or
17	catalysts tailored for pretreated feeds.
18	
19	The present invention includes a process for reverse staging to obtain high
20	conversion, selective hydrotreating and product selectivity in a
21	hydroprocessing reactor system including performing in a single reactor loop
22	a higher conversion or deeper treating processing in a top bed(s) of a reactor
23	or in the lead reactor in a series reactor loop and performing the general feed
24	processing in the reactor zones that follow.
25	IV. BRIEF DESCRIPTION OF THE DRAWINGS
26	
27	FIG. 1 depicts one embodiment of a flow diagram of the process of the
28	invention utilizing a common vessel for housing the different treatment zones
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FIG. 2 depicts an alternate embodiment of a flow diagram of the process of 1 2 the invention utilizing separate vessels for housing the different treatment 3 zones. 4 5 V. DETAILED DESCRIPTION OF PREFERRED EMBODIMENTS 6 7 A. Overview of the Process: Upper and Lower Reaction Zones 8 9 The invention is a method of reverse stage hydrotreating a hydrocarbon feed 10 to obtain high conversion, selective hydrotreating and product selectivity in a 11 hydroprocessing reactor system. The method includes passing a 12 hydrocarbon feed to a first hydrotreating zone, e.g., a denitrification and 13 desulfurization zone. In that zone, the hydrocarbon feed is contacted at 14 hydrotreating conditions, e.g., denitrification and desulfurization conditions, 15 with a hydrotreating catalyst, e.g., a denitrification and desulfurization 16 catalyst. After the contacting, a denitrification and desulfurization zone 17 effluent is recovered. 18 19 The denitrification and desulfurization zone effluent is then passed to a 20 purification/cooling zone (or termed a "NH<sub>3</sub> and H<sub>2</sub>S and cooling zone"). 21 Ammonia and H<sub>2</sub>S are removed, typically by water scrubbing. The effluent is 22 cooled by any conventional means, e.g., by heat exchanger. Recovered from

the purification/cooling zone is a hydrogen/light hydrocarbon stream from the top and a liquid stream containing dissolved gases from the bottom. The hydrogen/light hydrocarbon stream is optionally passed to a second H<sub>2</sub>S removal zone, typically using an amine adsorbent for H<sub>2</sub>S removal. The recovered effluent from the optional H<sub>2</sub>S removal zone is optionally passed to

a second hydrotreating zone, e.g., the hydrocracking zone discussed below.

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The liquid stream containing dissolved gases is passed to a separation zone. 1 2 Any conventional separation may be used, typically distillation. A light 3 product and other fractions selected from a liquid bottoms, one or more sideor mid-cuts, and mixtures thereof, are recovered. The other fractions, i.e., 4 5 liquid bottoms and/or one or more side- or mid-cuts, are passed to a second hydrotreating zone, e.g., a hydrocracking zone. There, at hydrocracking 6 conditions, the liquid bottoms and/or one or more side- or mid-cuts are 7 contacted with a hydrocracking catalyst. A hydrocracking zone effluent is 8 then recovered. The hydrocracking zone effluent is then passed to the first 9 hydrotreating zone, in one embodiment, a denitrification and desulfurization 10 11 zone. 12 The use of the two reaction zones can be varied in this invention. That is, the 13 first and second hydrotreating or reaction zone may each be a hydrocracking 14 zone or a denitrification and desulfurization zone. In one embodiment of the 15 invention, the lower zone which the fresh feed first contacts is a denitrification 16 and desulfurization zone. The upper feed is a hydrocracking zone. In 17 another embodiment, the reverse is true. Alternatively, each zone may both 18 be either a hydrocracking zone or each a denitrification and desulfurization 19 zone. Each may also be a combination or mixture of a hydrocracking zone 20 and a denitrification and desulfurization zone. 21 22 B. Advantages of Process of the Invention 23 24 The present invention provides a single reaction loop. This single reaction 25 loop method lowers costs as compared to the use of multiple reaction loops. 26 Yet, the single reaction loop of the invention maintains the advantages of 27 higher reaction rates or catalyst tailored for pretreated feeds of a multiple 28 reaction loop system. The present invention accomplishes the final 29

processing in the upp rraction zon or top b d or b ds of a reactor or

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reactors in series while performing the general feed processing in the lower reaction zones that follow.

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4 Another advantage of a series configuration rather than a parallel reactor 5 configuration for the initial conversion and one for the high conversion step is 6 that gas circulation is minimized thereby reducing both investment and 7 operating costs. The capital cost is lower due to smaller equipment and 8 piping. Operating costs are lower due to less compressor power to recirculate gas. The gas circulation is reduced relative to initial processing in 9 10 a separate loop or parallel reactor of the same loop because (a) the high 11 conversion effluent from the top reaction zone serves as a partial heat sink 12 and thereby reduces the quench requirements for the initial processing in the 13 zones which follow, (b) the unused hydrogen in the high conversion effluent from the top zone serves as a partial source of hydrogen for the initial 14 15 processing in the zones which follow, and (c) the high conversion effluent 16 from the top reaction zone helps to provide good distribution of the fresh feed 17 and hydrogen for reaction on the catalyst in the zones which follow. Thus, 18 the advantages of using a single loop are reduced investment cost and 19 operating costs by not duplicating similar pieces of equipment in two separate 20 loops, i.e., one for the initial processing and one for the high conversion step.

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Advantages of processing the pretreated hydrocarbon in upper reaction zone or top bed separate from the fresh feed include (a) the top bed catalysts are not contaminated with feed impurities, (b) the reaction rate in the top beds is not inhibited by substantial quantities of hydrotreating byproducts, e.g., NH<sub>3</sub> and H<sub>2</sub>S, and (c) hydrogen partial pressures are maximized for the finishing processes.

1 In an optional embodiment in the case of residuum processing, the present 2 process can also provide benefits in the lower reaction zones which includes 3 reduced pulsation tendency. 4 5 C. Feedstocks and Products 6 7 Feedstocks suitable for use in the invention and desired products obtained 8 include any conventional or known hydrocracking/hydroprocessing 9 feedstocks and products. The feedstocks and desired products for the 10 instant process include those disclosed in U.S. Patent Nos. 5,277,793; 11 5,232,577; 5,073,530; 4,430,203; and 4,404,088 which are incorporated 12 herein by reference. In one preferable embodiment, the hydrocarbon feed is 13 selected from a residuum, a vacuum gas oil, middle-distillates, and mixtures 14 thereof. 15 16 D. Reaction Conditions and Catalysts 17 18 Suitable hydrocracking and hydroprocessing catalysts and reaction conditions include any conventional or known catalysts and reaction 19 20 conditions. The catalysts and reaction conditions suitable for the instant 21 process include those disclosed in U.S. Patent Nos. 5,277,793; 5,232,577; 22 5,073,530; 4,430,203; and 4,404,088 which are incorporated herein by 23 reference. Where the reaction zone is a denitrification and/or desulfurization zone, the contacting occurs at denitrification and/or desulfurization 24 25 conditions. Where the reaction zone is a hydrocracking zone, the contacting 26 occurs at hydrocracking conditions. 27 28 When the above-described process is used to hydrotreat feedstocks to 29 remove sulfur and nitrogen impurities, the following process conditions will typically b used: reaction t mp rature, 400°F-900°F; pr ssure, 500 to 30

1	5000 psig; LHSV, 0.5 to 20; and overall hydrogen consumption 300 to
2	2000 scf per barrel of liquid hydrocarbon feed. The hydrotreating catalyst for
3	the beds will typically be a composite of a Group VI metal or compound
4	thereof, and a Group VIII metal or compound thereof supported on a porous
5	refractory base such as alumina. Examples of hydrotreating catalysts are
6	alumina supported cobalt-molybdenum, nickel sulfide, tungsten-nickel sulfide
7	cobalt molybdate and nickel molybdate.
8	
9	Correspondingly, when the process is used to hydrocrack feedstocks, the
10	following operating conditions will normally prevail: reaction temperature,
11	400°F-950°F; reaction pressure 500 to 5000 psig; LHSV, 0.1 to 15; and
12	hydrogen consumption 500 to 2500 scf per barrel of liquid hydrocarbon feed.
13	The hydrocracking catalysts used for the beds will typically be a Group VI,
14	Group VII, or Group VIII metal or oxides or sulfides thereof supported on a
15	porous refractory base such as silica or alumina. Examples of hydrocracking
16	catalysts are oxides or sulfides of Mo, W, V, and Cr supported on such
17	bases.
18	
19	Generally, where the reaction zone is a denitrification and/or desulfurization
20	zone, the catalyst is any catalyst which will catalyze denitrification and/or
21	desulfurization at denitrification and/or desulfurization conditions. Where the
22	reaction zone is a hydrocracking zone, the catalyst is any catalyst which will
23	catalyze hydrocracking at hydrocracking conditions.
24	
25	VI. <u>DETAILED DESCRIPTION OF THE DRAWINGS</u>
26	
27	Modifications of the process that is shown in the drawings and described in
28	this specification that are obvious to those of ordinary skill in the oil refinery
29	process art are intended to be within the scope of the invention.

#### A. Figure 1

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As illustrated in the flow diagram of FIG. 1, the catalytic reactions used in this 3 process are accomplished in two reaction zones 3 and 10. Vessel 2 houses 4 5 both reaction zones 3 and 10. The initial processing is carried out in the 6 second zone 10 and the high conversion processing carried out in the first 7 zone 3. The flow scheme optionally includes other features which are 8 common in hydroprocessing units such as preheating of liquid and gas feeds 9 to the reactors (preheaters not shown), NH<sub>3</sub> and H<sub>2</sub>S removal and effluent 10 cooling and separation zone 20, optional recycle gas purification zone 31, 11 and recirculation streams 30 and 32, and product separation and distillation zone 40. Liquid bottoms stream 50, and/or side- or mid-cut 52, from 12 13 distillation zone 40 are joined as stream 54. Stream 54 is passed to reaction 14 zone 3. Make-up hydrogen stream 60 is added to gas recirculation stream 32 (also termed "hydrogen/light hydrocarbon stream 30" or "H₂S removal zone 15 16 effluent 32"). Alternatively, make-up hydrogen stream 70 is added to feed 17 stream 1 instead of, or in addition to, adding make-up hydrogen to stream 32.

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Hydrocracking or deeper hydrotreating takes place in reaction zone 3 depending on the type of catalyst used in that zone. The effluent 65 from reaction zones passes to reaction zone 10. Fresh feed 1 is introduced at an intermediate point between reactor beds 3 and 10. It is processed in the presence of the effluent 65 from the upper reaction zone 3. Effluent 65 assists in distribution of feed stream 1 through reaction zone 10. Effluent 65 also acts as a heat-sink for the exothermic reaction in reaction zone 10.

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The effluent 15 from the lower zone 10 is treated for NH<sub>3</sub> and H<sub>2</sub>S removal in zone 20. Conventional methods, typically water washing, is utilized for the NH<sub>3</sub> and H<sub>2</sub>S removal. Zone 20 is also a cooling and separation zone producing a gas str am 30 and a liquid stream containing dissolved

gases 35. Conventional processing is used for the interrelationships of the 1 NH₃ and H₂S removal and cooling and separation processes in zone 20. 2 Zone 20 may include multiple units or sub-zones according to conventional 3 means for accomplishing the NH<sub>3</sub> and H<sub>2</sub>S removal and cooling and 4 separation. A hydrogen rich gas stream 32 is recycled back to the reactors 5 and then mixed with make-up hydrogen stream 60. Alternatively, or in 6 addition to mixing make-up hydrogen stream 60 with hydrogen rich gas 7 stream 32, make-up hydrogen stream 70 is mixed with oil feed stream 1. The 8 recycle gas in stream 30 is optionally purified, e.g., by amine adsorbent for 9 H<sub>2</sub>S removal, in zone 31 before recirculation to the reactors. The recycle gas 10 of stream 30 (or stream 32 if further purified in zone 31) is optionally fed to 11 stream 54 for feeding to first reaction zone 3 or is passed, as stream 34, to 12 feed stream 1 for feeding to second reaction zone 10. 13 14 15 B. Figure 2 16 The description of FIG. 2 is the same as for Figure 1 above, except for the 17 following differences. In Figure 1, a common vessel houses the reaction 18 zones. In Figure 2, separate vessels 2 and 9 house reaction zones 3 and 10, 19

respectively. In Figure 1, there is a recycle gas purification zone 31. In

20 21

Figure 2. this unit is omitted.

1		VII. <u>CLAIMS</u>
2		
3	WHA	IT IS CLAIMED IS:
4		
5	1.	A method of reverse stage hydrotreating a hydrocarbon feed to obtain
6		high conversion, selective hydrotreating and product selectivity in a
7		hydroprocessing reactor system, said method comprising:
8		a. Passing a hydrocarbon feed selected from a residuum, a vacuum
9		gas oil, middle distillate, and mixtures thereof to a denitrification
10		and desulfurization zone; contacting said hydrocarbon feed at a
11		temperature of about 400°F to about 900°F; a pressure of about
12		500 psig to about 5000 psig; a flow rate of about 0.5 LHSV to
13		about 20 LHSV; and an overall hydrogen consumption of about
14		300 to about 2000 scf per barrel of liquid hydrocarbon feed, with a
15		denitrification and desulfurization catalyst; and recovering a
16		denitrification and desulfurization zone effluent therefrom;
17		b. Passing said denitrification and desulfurization zone effluent to a
18		purification/cooling zone for removal of NH₃ and H₂S and cooling,
19		and recovering from said purification/cooling zone a hydrogen/ligh
20		hydrocarbon stream and a liquid stream containing dissolved
21		gases;
22		c. Passing said liquid stream containing dissolved gases to a
23		separation zone and recovering a light product, a liquid bottoms,
24		and at least one side-cut product therefrom;
25		d. Passing said liquid bottoms and said side-cut product and said
26		hydrogen/light hydrocarbon stream from step (b) to a
27		hydrocracking zon; contacting said liquid bottoms and said

1			side-cut product at a temperature of about 400°F to about 950°F; a
2			pressure of about 500 psig to about 5000 psig; a flow rate of about
3			0.1 LHSV to about 15 LHSV; and an overall hydrogen
4			consumption of about 500 to about 2500 scf per barrel of liquid
5			hydrocarbon feed, with a hydrocracking catalyst; and recovering a
6			hydrocracking zone effluent therefrom; and
7		e.	Passing said hydrocracking zone effluent to said denitrification
8			and desulfurization zone.
9	2.	A pre	ocess for reverse staging to obtain high conversion, selective
10		hydr	otreating and product selectivity in a hydroprocessing reactor
11		syste	em comprising performing in a single reactor loop a higher
12		conv	version or deeper treating processing in an upper reaction zone of a
13		reac	tor or in the lead reactor in a series reactor loop and performing the
14		gene	eral feed processing in the reaction zones that follow.
15	3.	The	process of claim 2 further comprising feeding make-up hydrogen to
16		said	upper reaction zone or the reaction zones that follow.
17	4.	The	process of claim 2 further comprising:
18		a.	Recovering an effluent from said reaction zones that follow and
19			passing said effluent to a cooling zone;
20		b.	Recovering from said cooling zone a hydrogen/light hydrocarbon
21			stream and a liquid hydrocarbon stream containing dissolved
22			gases;
23		C.	Passing said hydrogen/light hydrocarbon stream to said upper
24			reaction zone; and

- d. Passing said liquid stream containing dissolved gases to a
   separation zone.
- The process of claim 3 wherein said general feed is selected from a residuum, a vacuum gas oil, a middle distillate, and mixtures thereof; and further comprising passing said hydrogen/light hydrocarbon stream in step (b) to a H<sub>2</sub>S removal zone prior to passing to said upper reaction zone or lead reactor in step (c).
- 8 6. The process of claim 4 wherein said reactor zones that follow are
  9 hydrocracking zones and comprise a hydrocracking catalyst and
  10 wherein said hydrocracking zones have a temperature of about 400°F to
  11 about 950°F; a pressure of about 500 psig to about 5000 psig; a flow
  12 rate of about 0.1 LHSV to about 15 LHSV; and an overall hydrogen
  13 consumption of about 500 to about 2500 scf per barrel of liquid
  14 hydrocarbon feed.
- 15 The process of claim 5 wherein said reactor zones that follow are 7. denitrification and desulfurization zones and said process further 16 17 comprises contacting in said denitrification and desulfurization zones a 18 denitrification and desulfurization catalyst with a general feed selected from residuum, a vacuum gas oil, middle distillates, and mixtures 19 20 thereof, at a temperature of about 400°F to about 900°F; a pressure of about 500 psig to about 5000 psig; a flow rate of about 0.5 LHSV to 21 about 20 LHSV; and an overall hydrogen consumption of about 300 to 22 about 2000 scf per barrel of liquid hydrocarbon feed, and further 23 comprises recovering a denitrification and desulfurization zone effluent. 24

- The process of claim 4 wherein said reaction zones that follow are 1 8. hydrocracking zones and said process further comprises contacting in 2 said hydrocracking zones a hydrocracking catalyst with a general feed 3 selected from residuum, a vacuum gas oil, middle distillates, and 4 mixtures thereof, at a temperature of about 400°F to about 950°F; a 5 pressure of about 500 psig to about 5000 psig; a flow rate of about 6 0.1 LHSV to about 15 LHSV; and an overall hydrogen consumption of 7 about 500 to about 2500 scf per barrel of liquid hydrocarbon feed. 8
- The process of claim 7 wherein said upper reaction zone or lead reactor 9 9. is a hydrocracking zone and said process further comprises contacting 10 in said hydrocracking zones a hydrocracking catalyst with at least a 11 portion of said denitrification and desulfurization zone effluent, at a 12 temperature of about 400°F to about 950°F; a pressure of about 13 500 psig to about 5000 psig; a flow rate of about 0.1 LHSV to about 14 15 LHSV; and an overall hydrogen consumption of about 500 to about 15 2500 scf per barrel of liquid hydrocarbon feed. 16
- 17 10. The process of claim 4 wherein said upper reaction zone or lead reactor
  18 is a denitrification and desulfurization zone.
- 19 11. The process of claim 9 further comprising:
- a. Passing said denitrification and desulfurization zone effluent to a purification/cooling zone for removal of NH₃ and H₂S and cooling, and recovering from said purification/cooling zone a hydrogen/light hydrocarbon stream and a liquid stream containing dissolved gases;

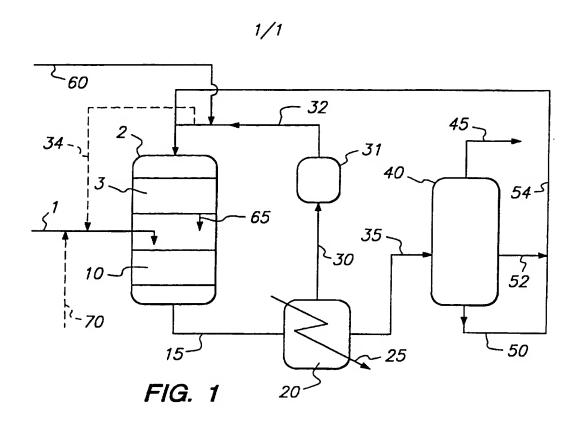
1		b.	Passing said liquid stream containing dissolved gases to a
2			separation zone and recovering a light product, a liquid bottoms,
3			and at least one side-cut product therefrom; and
4		C.	Passing said liquid bottoms and said side-cut product and said
5			hydrogen/light hydrocarbon stream from step (b) to said upper
6			reaction zone.
7	12.	A m	ethod of processing a hydrocarbon feed comprising:
8		a.	Passing a hydrocarbon feed to a second hydrotreating zone,
9			contacting at hydrotreating conditions said hydrocarbon feed with
10			a second hydrotreating catalyst, and recovering a second
11			hydrotreating zone effluent therefrom;
12		b.	Passing said hydrotreated product to a vapor-liquid separation
13			zone, and recovering therefrom a light product and other fractions
14			selected from a liquid bottoms, one or more middle cuts, and
15			mixtures thereof;
16		C.	Passing said other fractions to a first hydrotreating zone,
17			contacting at hydrotreating conditions said hydrocarbon feed with
18			a first hydrotreating catalyst, and recovering a first hydrotreating
19			zone effluent therefrom; and
20		d.	Passing said first hydrotreating zone effluent to said second
21			hydrotreating zone.
22	13.	The	process of claim 12 further comprising feeding make-up hydrogen
23		to s	aid second hydrotreating zone.
24	14.	The	process of claim 12 further comprising:

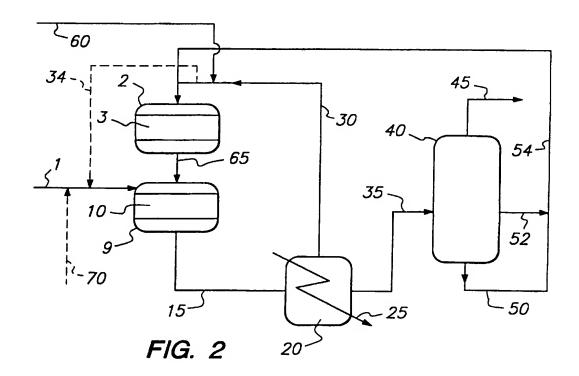
Passing said second hydrotreating zone effluent to a NH₃ and H₂S 1 a. 2 removal and cooling zone; Recovering from said NH3 and H2S removal and cooling zone a 3 b. hydrogen/light hydrocarbon stream and a liquid hydrocarbon 4 5 stream containing dissolved gases; Passing said hydrogen/light hydrocarbon stream to said first 6 C. 7 hydrotreating zone; and d. Passing said liquid hydrocarbon stream containing dissolved 8 gases to said vapor-liquid separation zone. 9 15. The process of claim 14 wherein said hydrocarbon feed is selected from 10 a residuum, a vacuum gas oil, middle-distillates, and mixtures thereof. 11 16. The process of claim 12 wherein said second hydrotreating zone is a 12 denitrification and desulfurization zone having a temperature of about 13 400°F to about 900°F; a pressure of about 500 psig to about 5000 psig; 14 a flow rate of about 0.5 LHSV to about 20 LHSV; and an overall 15 hydrogen consumption of about 300 to about 2000 scf per barrel of 16 liquid hydrocarbon feed, and wherein said second hydrotreating catalyst 17 comprises a denitrification and desulfurization catalyst. 18 17. The process of claim 16 wherein said second hydrotreating zone is a 19 20 hydrocracking zone.

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1	18.	The process of claim 12 wherein said first hydrotreating zone is a
2		hydrocracking zone having a temperature of about 400°F to about
3		950°F; a pressure of about 500 psig to about 5000 psig; a flow rate of
4		about 0.1 LHSV to about 15 LHSV; and an overall hydrogen
5		consumption of about 500 to about 2500 scf per barrel of liquid
6		hydrocarbon feed, and wherein said first hydrotreating catalyst
7		comprises a hydrocracking catalyst.
8	19.	The process of claim 12 wherein said first hydrotreating zone is a
9		denitrification and desulfurization zone.





# INTERNATIONAL SEARCH REPORT

In' ional Application No PCT/US 97/04270

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A. CLASS IPC 6	C10G65/12 C10G65/04		
According	to International Patent Classification (IPC) or to both national clas	safication and IPC	
B. FIELD	S SEARCHED		
Minimum (IPC 6	documentation searched. (classification system followed by classific ${\sf C10G}$	ation symbols)	
Documenta	ition searched other than minimum documentation to the extent tha	t such documents are included in the field	searched
Electronic	data base consulted during the international search (name of data b	ase and, where practical, search terms used	1)
C. DOCUN	MENTS CONSIDERED TO BE RELEVANT		
Category *	Citation of document, with indication, where appropriate, of the	relevant passages	Relevant to claim No.
γ	US 3 328 290 A (STANDARD OIL) 27 see the whole document	June 1967	1-19
Υ	US 4 059 503 A (THE LUMMUS COMPA November 1977 see the whole document	NY) 22	1-19
A	US 3 657 110 A (STANDARD OIL) 18 1972 see the whole document	April	1-19
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Furt	ner documents are listed in the continuation of box C.	X Patent family members are listed	in annex.
"A" docume consider of filing d "L" docume which is citation other n "P" docume	ent which may throw doubts on priority claim(s) or is cited to establish the publication date of another is or other special reason (as specified) in or other special reason (as specified) in the first referring to an oral disclosure, use, exhibition or	"T" later document published after the in or priority date and not in conflict we cited to understand the principle or timention."  "X" document of particular relevance; the cannot be considered novel or cannot involve an inventive step when the d."  "Y" document of particular relevance; the cannot be considered to involve an indocument is combined with one or ments, such combination being obvious the art.  "&" document member of the same paten.	nth the application but heavy underlying the claimed invention to considered to occurrent is taken alone claimed invention exercises when the love other such docurrent to a person skilled
	sectual completion of the international search  July 1997	Date of mailing of the international s	
	nailing address of the ISA  European Patent Office, P.B. 5818 Patentiaan 2  NL - 2280 HV Rijswijk  Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  Fax: (+31-70) 340-3016	Authorized officer Michiels, P	

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information on patent family members

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